

# Al-Mg-Sc-Zr (Aluminum-Magnesium-Scandium-Zirconium)

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Recent work on this quaternary system pertains to partial isothermal sections for Al-rich alloys at 500 and 430 °C [1992Fri, 2004Zen].

## Binary and Ternary Subsystems

For brief descriptions of the phases in the Al-Mg, Al-Sc and Mg-Sc systems, see the Al-Mg-Sc update in this review. There are a number of intermediate phases in the Al-Zr system. The only phase of interest in the results on Al-rich alloys reviewed here is  $ZrAl_3$  ( $D0_{23}$ -type tetragonal). There are no intermediate phases in the Mg-Zr and Sc-Zr systems.

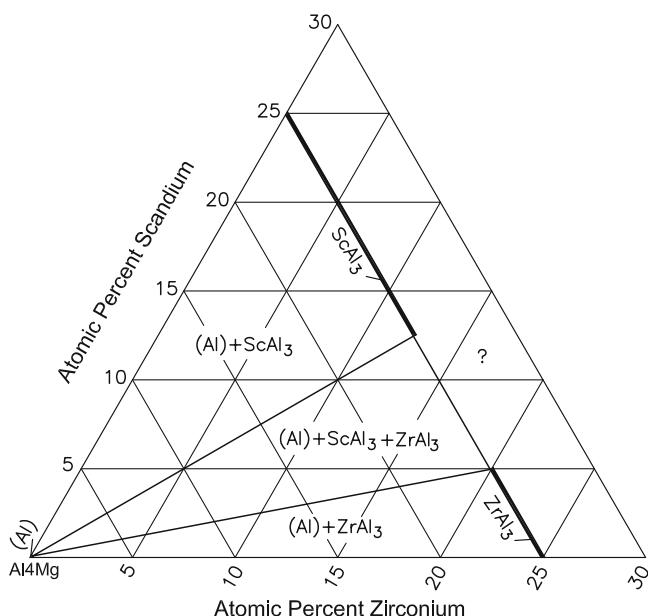
Updates of the Al-Mg-Sc system appear in this issue and in [2007Rag]. [2004Zen] reviewed briefly the phase diagram data on the Al-Mg-Zr and Al-Sc-Zr systems.

## Quaternary Phase Equilibria

[1991Rok] employed differential thermal analysis and a sedimentation method to determine the temperature profile of the liquidus surface at a constant 6 mass% Mg. Here, after equilibration of a liquid of known composition at a given temperature, a small quantity of the liquid was sucked out from the surface of the melt through a syringe. The agreement of the composition of this liquid with the known composition of the alloy indicated a single phase. A decrease in the Sc/Zr content indicated precipitation, as the precipitates sink to the bottom of the crucible. The liquidus temperatures were thus bracketed. An equation was obtained by [1991Rok] by regression analysis for the temperature  $T$  of the liquidus, as a function of Sc and Zr contents in mass% at a constant Mg of 6 mass%:

$$T(\text{°C}) = 602.24 + 211.15[\text{Sc}] + 642.69[\text{Zr}] - 78.14[\text{Sc}]^2 - 697.43[\text{Zr}]^2$$

With starting metals of 99.99% Al, 99.96% Mg, 99.91% Sc and 99.90% Zr, [1992Fri] induction-melted about 40 alloy compositions with a constant Mg content of 6 mass%, Sc and Zr up to 3 and 4 mass% respectively. The alloys were annealed at 500 °C for 200 h and quenched in water. The phase equilibria were studied by metallography, x-ray powder diffraction (XRD) and x-ray spectral microanalysis. The isothermal section at 500 °C and at 6 mass% Mg shows that (Al) dissolves <0.1 mass% Sc or Mg and forms tie-lines with  $\text{ScAl}_3$  and  $\text{ZrAl}_3$ . [1992Fri] found that the maximum solubility of Zr in  $\text{ScAl}_3$  is 23 mass% and that of Sc in  $\text{ZrAl}_3$  is 4 mass%.



**Fig. 1** Al-Mg-Sc-Zr isothermal section at 430 °C and at 4 at.% Mg [2004Zen]

With starting materials of 99.99% Al, 99.99% Mg, Al-1.77 mass% Sc, and Al-4.42 mass% Zr, [2004Zen] induction-melted under Ar atm 18 quaternary alloys with Sc and Zr up to 2.0 and 2.5 at.% respectively and a constant 4 at.% Mg. The alloys were annealed at 430 °C for 360 h and quenched in water. The phase equilibria were studied by optical and electron metallography, XRD and EPMA. The isothermal section at 430 °C and at 4 at.% Mg constructed by [2004Zen] is redrawn in Fig. 1 to comply with the binary data. It is similar to that found by [1992Fri] at 500 °C and 6 mass% Mg. The solubility limits of Zr and Sc in  $\text{ScAl}_3$  and  $\text{ZrAl}_3$  respectively are given by  $\text{Sc}_{0.5}\text{Zr}_{0.5}\text{Al}_3$  and  $\text{Zr}_{0.8}\text{Sc}_{0.2}\text{Al}_3$  [2004Zen].

## References

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